All dimensions are nominal. Dimensions in [ ] are in millimeters.

## Service Conditions



## Valve Requirements

Required Fail Position: $\qquad$
Body
Material:

Size: $\qquad$ End Connections: $\qquad$
Trim Check one $\quad \square$ Modified Linear $\quad \square$ Equal Percentage
Material: $\qquad$ Close-off Class: $\qquad$
Additional Requirements: $\qquad$

## Actuator Requirements



| Name: | Company: |
| :--- | :--- |
| Date: |  |
| Project Name: | P.O. Number: |

## VALVE SELECTION

The proper sizing of a valve is one of the most important factors in the ability of a loop to maintain control. A valve that is too small is not able to provide the desired capacity during peak load conditions, while a valve that is too large will tend to overshoot the control point and operate with the valve plug too close to the seat, resulting in undue wear of the plug and seat.

## Valve Coefficient ( $C_{v}$ )

The valve coefficient $\left(\mathbf{C}_{\mathbf{v}}\right)$ is mathematically determined through an evaluation of the system service conditions. This factor can be used to select a valve body of the appropriate port size. In almost all cases, the valve should be of a smaller size than the pipeline into which it will be installed. To avoid undue wear, a valve body of the smallest possible port size should be selected; however, the valve should never be less than half the pipeline size, as this will cause extreme mechanical stress to the pipeline.

## Service Conditions

The specifier should be knowledgeable of the service conditions of the application in order to properly determine the actuator and valve requirements.

## Medium

The composition of the fluid passing through the valve.

## Temperature ( T )

The temperature of the medium passing through the valve. This measurement is required to properly specify the materials used to manufacture the valve.

## Flow (q or W)

The volume of fluid passed through the valve as required by the particular application. Flow is usually expressed as either gallons per minute (q), or pounds per hour (W). Water and other liquids are usually measured in gallons per minute, while steam and other gases are usually measured in pounds per hour. This measurement is required to correctly determine the valve coefficient ( $\mathbf{C}_{\mathbf{v}}$ ).
Inlet Pressure (Upstream Pressure or $\mathrm{P}_{1}$ )
The pressure (psia) of the medium flowing into the valve body. This measurement is required to correctly determine the valve coefficient ( $\mathbf{C}_{\mathbf{v}}$ ) and valve close-off capability.

## Outlet Pressure (Downstream Pressure or P2)

The pressure (psia) of the medium flowing through a fully opened valve to the process. The outlet pressure from the valve is determined by the process or equipment that is being fed by the valve, and is not caused by the valve itself. This measurement is required to correctly determine the valve coefficient $\left(\mathbf{C}_{\mathbf{v}}\right)$ and valve close-off capability.
Differential Pressure (Pressure Drop or $\Delta \mathbf{P}$ )
The difference between the inlet and outlet pressures $\left(\mathbf{P}_{1}-\mathbf{P}_{\mathbf{2}}\right)$. This measurement is required to correctly determine the valve coefficient ( $\mathbf{C}_{\mathbf{v}}$ ) and valve close-off capability.

## Valve Sizing Differential Pressure:

The differential pressure (psid) for valve sizing is determined with the valve full open. This pressure drop, along with the required flow rate, is used to determine the required $\mathbf{C}_{\mathbf{v}}$ to aid in the selection of the proper control valve.

## Close-Off Differential Pressure:

The differential pressure (psid) for valve "close-off" is determined with the valve fully closed. Usually, in most common applications, with the valve closed the outlet pressure will be zero (0) psig and as such the pressure drop will be equal to the Inlet Pressure. In some applications there may be residual back pressure in the downstream system (such as filling a pressurized tank) that will cause the Outlet Pressure to be a value greater than zero, which in turn reduces the value of the expected differential pressure.

## Example:

Valve B73 (on page 190) has a maximum "Close-Off Pressure" allowance of 65 psid. If this valve is used to control the flow into an open tank, the closed valve outlet pressure will be zero. As such the maximum inlet pressure that the actuator can close this valve against is 65 psig .
( 65 psid rating + zero outlet pressure $=65$ psig inlet pressure).
If however this same valve B73 is used to control the flow into a closed pressurized tank (pressurized to 25 psig) then the maximum inlet pressure that the actuator can close this valve against is 90 psig. ( 65 psid rating +25 psig back pressure $=90$ psig inlet pressure).

Since this 90 psig is less than the body rating of 125 psig this valve would be acceptable for this service.

## VaLVE SELECTION

## Other Considerations

- Specific Gravity - The ratio between the weight of the flow medium at the flow temperature and that of a defined standard substance (water or air). The specific gravity may be required to correctly determine the valve coefficient ( $\mathbf{C}_{\mathbf{v}}$ ).

$$
\begin{array}{ll}
\text { Liquids }\left(\mathbf{G}_{\mathbf{f}}\right) & \text { water }=1.0 @ 39^{\circ} \mathrm{F}\left(4^{\circ} \mathrm{C}\right) \\
\text { Gases }\left(\mathbf{G}_{\mathbf{g}}\right) & \text { air }=1.0 @ 60^{\circ} \mathrm{F}\left(18^{\circ} \mathrm{C}\right) \text { and } 14.7 \text { psia }
\end{array}
$$

- Viscosity - The degree of thickness of a liquid. Extremely thick process media can create high friction as it passes through the valve. In most instances a sizing correction factor is not required. Please consult the factory when the flow medium is of a viscosity of 40 centistokes or greater.
- Steam Superheat - The number of degrees Fahrenheit ( $\mathbf{T}_{\mathbf{s h}}$ ) above the saturation temperature of steam at a given pressure. Superheated steam is created when saturated steam is further heated from another source after leaving the water from which it is formed. This measurement is required to correctly determine the valve coefficient $\left(\mathbf{C}_{\mathbf{v}}\right)$.


## Valve Sizing Equations

The following formulas can be used to determine the $\mathrm{C}_{V}$ requirement for a specific set of service conditions,
where:

```
C
G}\mp@subsup{\mathbf{f}}{\mathbf{f}}{=}\mathrm{ liquid specific gravity at flow temperature
    (water = 1.0)
    G}\mp@subsup{\mathbf{g}}{\mathbf{g}}{=}\mathrm{ gas specific gravity (air = 1.0)
    P}\mp@subsup{\mathbf{P}}{\mathbf{1}}{=}\mathrm{ inlet pressure (psia)
    P2 = outlet pressure (psia)
    \Delta\mathbf{P}= inlet pressure minus outlet pressure (psi)
    q = liquid flow in gallons per minute (gpm)
    T
    W = gas flow in pounds per hour (pph)
```


## Cavitation

Water and Other Liquids
Cavitation takes place when the pressure through a valve drops to or below the vapor pressure of a liquid, causing it to vaporize and rapidly expand in gas form. Vapor bubbles flow downstream where the fluid velocity decreases and the surrounding pressure increases. The vapor bubbles then collapse or implode, causing sudden condensation and producing shock waves that may result in excessive noise, vibration, erosion or mechanical damage to valve and/or piping. In most liquid applications, the outlet pressure (psia) should be no less than one-third the inlet pressure (psia). Where extremely large differential pressures are required, the use of multiple valves in series will reduce the possibility of cavitation.

## Water

where:
$\mathbf{q}=$ liquid flow in gallons per minute (gpm)
$\boldsymbol{\Delta} \mathbf{P}=$ inlet pressure minus outlet pressure (psi)

$$
\mathbf{C}_{\mathbf{v}}=\frac{\mathbf{q}}{\sqrt{\Delta \mathbf{P}}}
$$

example:

$$
\begin{aligned}
& \text { medium }=\text { water } \\
& \mathbf{q}=160 \text { U.S. gallons per minute } \\
& \mathbf{\Delta P}=25[100 \text { psia inlet }-75 \text { psia outlet }] \\
& \mathbf{C}_{\mathbf{v}}=\frac{160}{\sqrt{25}} \text { or } \mathbf{C}_{\mathbf{v}}=\frac{160}{5} \text { or } \mathbf{C}_{\mathbf{v}}=32
\end{aligned}
$$

## Technical Information

## VALVE SELECTION

## Saturated Steam

where:
$\mathbf{W}=$ gas flow in pounds per hour (pph)
$\mathbf{P}_{1}=$ inlet pressure (psia)
$P_{2}=$ outlet pressure (psia)
$\Delta \mathbf{P}=$ inlet pressure minus outlet pressure (psi)

$$
C_{\mathbf{v}}=\frac{\mathbf{w}}{2.1 \sqrt{\Delta P\left(P_{1}+P_{2}\right)}}
$$

example:
medium $=$ saturated steam
$\mathbf{W}=4000$ pph
$\mathbf{P}_{\mathbf{1}}=100$ psia
$\mathbf{P}_{\mathbf{2}}=75 \mathrm{psia}$
$\Delta \mathbf{P}=25$ [100 psia inlet -75 psia outlet]
$\mathbf{C}_{\mathbf{v}}=\frac{4000}{2.1 \sqrt{25(100+75)}}$
$C_{V}=\frac{4000}{138.9}$
$C_{v}=28.8$

## Choked Flow (Critical Drop)

## Steam and Other Gases

When $P_{2}$ is less than $1 / 2 P_{1}$, set $P_{2}$ equal to $1 / 2 P_{1}$ in the appropriate sizing equation for steam or gases.

Steam, as are all gases, is a compressible fluid. The maximum velocity of the steam or gas through the valve is limited to the speed of sound. When the outlet pressure (psia) is equal to one-half (or less) of the inlet pressure (psia), the fluid velocity through the valve reaches the speed of sound, and flow cannot be further increased by a reduced outlet pressure. This is known as a choked flow condition. The pressure drop under these conditions is known as critical drop.
example:
medium = saturated steam

$$
\begin{aligned}
& \mathrm{W}=4000 \text { pph } \\
& P_{1}=100 \text { psia } \\
& P_{2}=35 \text { psia (actual) } \\
& \Delta P=65 \text { psia }
\end{aligned}
$$

Since the outlet pressure is less than $1 / 2$ of the inlet pressure, choked flow will occur. Set $\mathbf{P}_{2}$ to equal $1 / 2$ of $\mathbf{P}_{1}$. Use this revised $P_{2}$ in the normal sizing formulae.

| $P_{2}=P_{1} / 2$ | $P_{2}=100 / 2$ | $P_{2}=50$ |
| :--- | :--- | :--- |
| $\Delta P=P_{1}-P_{2} \quad \Delta P=P_{1}-P_{1} / 2$ | $\Delta P=100-50$ | $\Delta P=50$ |
| $C_{v}=\frac{W}{2.1 \sqrt{\Delta P\left(P_{1}+P_{2}\right)}}$ |  |  |
| $C_{v}=\frac{4000}{2.1 \sqrt{50(100+50)}}$ |  |  |
| $C_{v}=\frac{4000}{181.9}$ |  |  |
| $C_{v}=22$ |  |  |

$C_{v}=22$

Steam is perfectly transparent，colorless，dry and invisible．When it comes in contact with air，it partially condenses and forms a visible mist，or wet steam．Wet steam has the same temperature as dry steam contained under the same pressure．

Steam in its most common state is known as saturated steam．Its temperature is the same as that of the water from which it is formed and is dependent on the pressure under which it is contained．Superheated steam is created when saturated steam is further heated from another source after leaving the water from which it is formed．

| Pressure | Temperature | Pressure | Temperature | Latent Heat |
| :---: | :---: | :---: | :---: | :---: |
| psig | ${ }^{\circ} \mathrm{F}$ | kPag | ${ }^{\circ} \mathrm{C}$ | BTU／lb |
| 0 | 212 | 0 | 100 | 970 |
| 5 | 227 | 34 | 108 | 961 |
| 10 | 239 | 69 | 115 | 953 |
| 15 | 250 | 103 | 121 | 946 |
| 20 | 259 | 138 | 126 | 939 |
| 25 | 267 | 172 | 130 | 934 |
| 30 | 274 | 207 | 134 | 929 |
| 35 | 281 | 241 | 138 | 924 |
| 40 | 287 | 276 | 142 | 920 |
| 45 | 292 | 310 | 145 | 916 |
| 50 | 298 | 345 | 148 | 912 |
| 60 | 307 | 414 | 153 | 905 |
| 70 | 316 | 483 | 158 | 898 |
| 80 | 324 | 552 | 162 | 892 |
| 90 | 331 | 621 | 166 | 886 |
| 100 | 338 | 689 | 170 | 881 |
| 110 | 344 | 758 | 173 | 875 |
| 120 | 350 | 827 | 177 | 871 |
| 130 | 356 | 896 | 180 | 866 |
| 140 | 361 | 965 | 183 | 861 |
| 150 | 366 | 1034 | 185 | 857 |
| 175 | 377 | 1207 | 192 | 847 |
| 200 | 388 | 1379 | 198 | 837 |
| 225 | 397 | 1551 | 203 | 828 |
| 250 | 406 | 1724 | 208 | 820 |

Adjusted to Sea Level－ 14.696 psia（ 760 mm Hg ）

## Technical Information

## STEAM PROPERTIES

- One cubic foot of water will become 1646 cubic feet of steam when evaporated at zero psi gauge pressure and a temperature of $212^{\circ} \mathrm{F}$.
- One cubic foot of steam weighs 0.03732 pounds, and one pound of steam occupies 26.796 cubic feet at zero psi gauge pressure and a temperature of $212^{\circ} \mathrm{F}$.
- One cubic foot of dry air weighs 0.08073 pounds, and one pound of dry air occupies 12.387 cubic feet at zero psi gauge pressure and a temperature of $0^{\circ} \mathrm{F}$.
- The latent heat created from the vaporization of water to steam is:

970 BTU per pound @ 14.7 psia 889 BTU per pound @ 100 psia

- One British Thermal Unit (BTU) is the amount of heat required to raise the temperature of one pound of water by one degree Fahrenheit,

1 Gal. (U.S.) $=0.1337 \mathrm{ft}^{3}$
1 Gal. (U.S.) water $=8.337 \mathrm{lbs}$
$1 \mathrm{ft}^{3}$ water $=62.364 \mathrm{lbs}$

## Steam Required to Heat Water

## According to Temperature Rise and Gallons per Hour

| U.S. Gallons of Water Heated per Hour (for fuel oil - multiply pounds per hour listed by 0.5) |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 25 | 50 | 75 | 100 | 150 | 200 | 300 | 400 | 500 | 750 | 1000 | 1500 | 2000 | 3000 | 4000 | 5000 | 7500 | 10000 |
| Temperature Rise ( ${ }^{\circ} \mathrm{F}$ ) | Pounds of Steam per Hour Required |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| 10 | - | - | - | - | - | 17 | 25 | 33 | 42 | 63 | 83 | 120 | 167 | 250 | 330 | 420 | 620 | 830 |
| 20 | - | - | - | - | 25 | 33 | 50 | 67 | 83 | 125 | 167 | 250 | 330 | 500 | 670 | 830 | 1250 | 1670 |
| 30 | - | - | - | 25 | 37 | 50 | 75 | 100 | 125 | 190 | 250 | 370 | 500 | 750 | 1000 | 1250 | 1900 | 2500 |
| 40 | - | - | 25 | 33 | 50 | 66 | 100 | 130 | 170 | 250 | 330 | 500 | 660 | 1000 | 1330 | 1700 | 2500 | 3300 |
| 50 | - | 21 | 31 | 42 | 63 | 84 | 125 | 170 | 210 | 310 | 420 | 630 | 840 | 1250 | 1680 | 2100 | 3100 | 4200 |
| 60 | 12 | 25 | 37 | 50 | 75 | 100 | 150 | 200 | 250 | 370 | 500 | 750 | 1000 | 1500 | 2000 | 2500 | 3700 | 5000 |
| 80 | 16 | 33 | 50 | 67 | 100 | 130 | 200 | 270 | 330 | 500 | 670 | 1000 | 1340 | 2000 | 2700 | 3300 | 5000 | 6700 |
| 100 | 21 | 42 | 63 | 83 | 120 | 170 | 250 | 330 | 420 | 630 | 830 | 1250 | 1700 | 2500 | 3300 | 4200 | 6300 | 8300 |
| 120 | 25 | 50 | 75 | 100 | 150 | 200 | 300 | 400 | 500 | 750 | 1000 | 1500 | 2000 | 3000 | 4000 | 5000 | 7500 | 10000 |
| 140 | 29 | 58 | 88 | 117 | 175 | 230 | 350 | 470 | 580 | 880 | 1170 | 1750 | 2340 | 3500 | 4700 | 5800 | 8800 | 11700 |
| 160 | 33 | 66 | 100 | 133 | 200 | 270 | 400 | 530 | 660 | 1000 | 1330 | 2000 | 2700 | 4000 | 5300 | 6600 | 10000 | 13300 |
| 180 | 37 | 75 | 113 | 150 | 225 | 300 | 450 | 600 | 750 | 1125 | 1500 | 2200 | 3050 | 4500 | 5950 | 7500 | 11300 | 14950 |
| 200 | 42 | 84 | 126 | 165 | 250 | 330 | 500 | 660 | 840 | 1260 | 1660 | 2500 | 3400 | 5000 | 6600 | 8300 | 12600 | 16600 |

Technical Information PRESSURE \&
Using the Table

| 1. Find the units you wish to convert FROM in the left hand column. 2. Find the units you wish to convert TO in the top row. <br> 3. Insert the multiplier shown at the intersection into the following formula: FROM units $\times$ MULTIPLIER $=$ TO units <br> Example: $100 \mathrm{psi} \times 6.894757=689.475 \mathrm{kPa}$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| TROM | PSI | in H 2 O | mm H20 | cm H2O | 0z/in2 | mbar | bar | mm Hg | cm Hg | in Hg | kg/cm2 | kPa | MPa | ft H2O | m H2O | atm |
| psi | 1 | 27.68068 | 703.1 | 70.308927 | 16 | 68.95 | 0.06894757 | 51.71486 | 5.171486 | 2.03602 | 0.070306958 | 6.894757 | 0.0069 | 2.306723 | 0.70308927 | 0.0680460 |
| in $\mathrm{H}_{2} \mathrm{O}$ | 0.03612628 | 1 | 25.4 | 2.54 | 0.578020 | 2.488 | 0.00249 | 0.0735539 | 0.187 | 0.0735539 | 0.00254219 | 0.2490819 | 0.00025 | 0.08333 | 0.0254 | 0.00245825 |
| mm H20 | 0.001422 | 0.0394 | 1 | 0.1 | 0.0227 | 0.098 | 0.000098 | 0.0735 | 0.00735 | 0.00289 | 0.0001 | 0.0098 | 0.00001 | 0.00328084 | 0.001 | 0.000097 |
| cm $\mathrm{H}_{2} \mathrm{O}$ | 0.0142229 | 0.3937 | 10 | 1 | 0.227566 | 0.98 | 0.000980634 | 0.7355372 | 0.0735 | 0.0289581 | 0.00099997 | 0.980634 | 0.0001 | 0.032808 | 0.01 | 0.000967814 |
| 0z/in2 | 0.0625 | 1.73004 | 43.943 | 4.394308 | 1 | 4.31 | 0.004309223 | 3.23218 | 0.323 | 0.12725125 | 0.04394308 | 0.4309223 | 0.00043 | 0.14417 | 0.04394308 | 0.004252875 |
| mbar | 0.0145 | 0.4012 | 10.20 | 1.020 | 0.2321 | 1 | 0.001 | 0.75 | 0.075 | 0.0295 | 0.00102 | 0.1 | 0.0001 | 0.03345622 | 0.00101975 | 0.000987 |
| bar | 14.5038 | 401.8596 | 10,197 | 1019.7466 | 232.0608 | 1000 | 1 | 750.0626 | 75 | 29.53 | 1.019716 | 100 | 0.1 | 33.4833 | 10.197466 | 0.986923 |
| mm Hg | 0.0193368 | 0.535255 | 13.60 | 1.359554 | 0.3093888 | 1.333 | 0.001333225 | 1 | 0.1 | 0.039370079 | 0.00135951 | 0.1333225 | 0.000133 | 0.0446046 | 0.01359554 | 0.0013157895 |
| cm Hg | 0.1934 | 5.358 | 136.0 | 13.60 | 3.10 | 13.33 | 0.01333 | 10 | 1 | 0.394 | 0.0136 | 1.333 | 0.00133 | 0.44604625 | 0.13595509 | 0.01316 |
| in Hg | 0.4911542 | 13.595484 | 345.3 | 34.53253 | 7.85847 | 33.86 | 0.03386389 | 25.4 | 2.54 | 1 | 0.0345316 | 3.386389 | 0.00339 | 1.132957 | 0.3453253 | 0.0334211 |
| kg/cm ${ }^{2}$ | 14.223343 | 393.711806 | 10,000.3 | 1000.028 | 227.57349 | 980.7 | 0.98066494 | 735.5588 | 73.56 | 28.95901 | 1 | 98.066494 | 0.0981 | 32.809312 | 10.00028 | 0.967841598 |
| kPa | 0.1450377 | 4.014742 | 101.97 | 10.19745 | 2.320603 | 10 | 0.01 | 7.500610 | 0.75 | 0.2952997 | 0.01019716 | 1 | 0.001 | 0.3345618 | 0.1019745 | 0.009869235 |
| MPa | 145.04 | 4019 | 101,975 | 10,197 | 2321 | 10,000 | 10 | 7500 | 750 | 295.3 | 10.2 | 1000 | 1 | 334.56218 | 101.9748043 | 9.869 |
| ft $\mathrm{H}_{2} \mathrm{O}$ | 0.433515 | 12 | 304.80 | 30.48 | 6.93624 | 29.88981 | 0.02988981 | 22.4192 | 2.24192 | 0.882646 | 0.03047912 | 2.988981 | 0.002988981 | 1 | 0.3048 | 0.02949896 |
| m $\mathrm{H}_{2} \mathrm{O}$ | 1.42229 | 39.370079 | 1000 | 100 | 22.7566 | 980.66494 | 0.98066494 | 73.55372 | 7.35537 | 2.89581 | 0.099997 | 9.8063439 | 0.0098063439 | 3.2808399 | 1 | 0.0967814 |
| atm | 14.696 | 406.794 | 10,333 | 1033.2633 | 235.136 | 1013 | 1.0132535 | 760 | 76 | 29.9213 | 1.033231 | 101.32535 | 0.1013 | 33.8995 | 10.332633 | 1 |

Temperature Conversion
At sea level:
Water boils at $212^{\circ} \mathrm{F}, 100^{\circ} \mathrm{C}$ and $80^{\circ} \mathrm{R}$ Water freezes at $32^{\circ} \mathrm{F}, 0^{\circ} \mathrm{C}$ and $0^{\circ} \mathrm{R}$

